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Chapter 7 Periodic Review Answer Key

Properties of the Elements

Chapter 7 - Periodic

Properties of the Elements:

Part 1 of 11

Quantum Numbers, Atomic
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Configurations

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The Periodic Table: Atomic
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and Electronegativity

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(25) Chemical formula.
indicates the relative

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number of atoms of each kind
in a chemical compound.
Reveals the number of atoms
of each element contained in
a single molecule ...

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CHAPTER 7 REVIEW Chemical
Formulas and Chemical
Compounds SECTION 1 SHORT
ANSWER Answer the following
questions in the space
provided. 1. c In a Stock
system name such as

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iron(III) sulfate, the Roman numeral tells us (a) how many atoms of Fe are in one formula unit. (b) how many sulfate ions can be attached to the iron atom. (c) the charge on each Fe ion.

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*7 Chemical Formulas and
Chemical Compounds*

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CHAPTER 7 REVIEW Chemical
Formulas and Chemical
Compounds MIXED REVIEW SHORT
ANSWER Answer the following
questions in the space

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provided. 1. Write formulas
for the following compounds:
CuCO₃ a. copper(II)
carbonate Na₂SO₃ b. sodium
sulfite (NH₄)₃PO₄ c.
ammonium phosphate SnS₂ d.
tin(IV) sulfide HNO₂ e.
nitrous acid 2.

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Arrangement of Electrons in Atoms chapter of this Holt McDougal Modern Chemistry Companion Course helps students learn the essential lessons associated with how electrons are arranged in atoms.

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CHAPTER 5 REVIEW The
Periodic Law SECTION 1 SHORT

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ANSWER Answer the following questions in the space provided. 1. c In the modern periodic table, elements are ordered (a) according to decreasing atomic mass. (b) according to Mendeleev's original design. (c)

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According to increasing
atomic number. (d) based on
when they were discovered.
2. d Mendeleev noticed that
certain similarities in the
...

5 The Periodic Law

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3d104s1 PROBLEMS Write the answer on the line to the left Show all your work in the space provided 9 1 1012 m What is the wavelength of light that has a frequency of 3×10^4 Hz in a vacuum ...

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