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I have used Kasap's 3rd edition of "Principles of Electronic Materials and Devices" as a course textbook for the 2nd year "Materials Physics" course (in Department of Materials Science & Engineering (MSE) at University of Toronto (UoT)).

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Solution From Coulomb's law of electrostatic attraction we know that the PE between two charges Q1 and Q2 separated by a distance r is given by  $Q_1 Q_2 / (4\pi\epsilon_0 r)$  First we consider the interaction between Na<sup>+</sup> ion and 6Cl<sup>-</sup> ions at distance r Applying Coulomb's law we have  $Q_1 Q_2 / (4\pi\epsilon_0 r) = 6e \cdot PE = = 4\pi\epsilon_0 r / 4\pi\epsilon_0 r$  Similarly, we now consider 12 Na<sup>+</sup> ions as second nearest neighbors at a distance r  $Q_1 Q_2 / (4\pi\epsilon_0 r) = 12e \cdot PE = = 4\pi\epsilon_0 r / 4\pi\epsilon_0 r$  and Cl ...

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Ernest Rutherford (1871-1937) 1.37 fSolutions to Principles of Electronic Materials and Devices: 2nd Edition (Summer 2001) Chapter 2 Second Edition ( 2001 McGraw-Hill) Chapter 2 2.1 Electrical conduction Na is a monovalent metal (BCC) with a density of 0.9712 g cm<sup>-3</sup>. Its atomic mass is 22.99 g mol<sup>-1</sup>.

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C0078028183 SM - Solutions Manual to Principles of Electronic Materials and Devices Safa Kasap. Solutions Manual to Principles of Electronic Materials and Devices Safa Kasap Chapter 2. University. Shahjalal University of Science and Technology. Course. Electrical Properties of Materials (EEE 327) Academic year. 2017/2018

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